

CBCS SCHEME

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15NT35

Third Semester B.E. Degree Examination, June/July 2018 Physical and Chemical Principles of Nano Technology

Time: 3 hrs.

Max. Marks: 80

Note: Answer any FIVE full questions, choosing ONE full question from each module.

Module-1

- 1
- Define quantum mechanics. State uncertainty principle with an example and add a note on complementarity. (05 Marks)
 - State and explain dual nature of matter by de Broglie. (03 Marks)
 - Derive an expression for uncertainty principle with help of localization experiment. (04 Marks)
 - Enumerate Planck's hypotheses. (04 Marks)

OR

- 2
- Give a brief note on valence Bond theory and its applications. (05 Marks)
 - Write a brief note on molecular orbital theory and its application. (06 Marks)
 - Write a short note on computational chemistry and name few applications. (05 Marks)

Module-2

- 3
- Define thermodynamics. Write a note on importance and limitations of thermodynamics. (08 Marks)
 - State and explain the first law of thermodynamics with mathematical expressions. (08 Marks)

OR

- 4
- Define entropy. Give a brief note on the illustration of the concept of entropy. (06 Marks)
 - Explain the concept of heat capacity at constant volume and constant pressure. (06 Marks)
 - What is spontaneous process? Write the criteria for spontaneity. (04 Marks)

Module-3

- 5
- Explain Debye theory of molar heat capacity and limitations. (08 Marks)
 - Derive Kronig-Penny model. (08 Marks)

OR

- 6
- Explain classical theory of molar heat capacity and limitations. (06 Marks)
 - Distinguish between metal, insulator and semiconductor. (06 Marks)
 - Discuss the concept of Lattice vibrations and thermal heat capacity. (04 Marks)

Module-4

- 7 a. Explain the following :
- i) Intrinsic semiconductors (08 Marks)
 - ii) Extrinsic semiconductors. (04 Marks)
- b. Discuss the concept of tunnelling. (04 Marks)
- c. Explain brief about classical and quantum tunnelling.

OR

- 8 a. Explain briefly P-N junction semiconductor diode and give its advantages. (10 Marks)
- b. Write a note on rectification. (06 Marks)

Module-5

- 9 a. Discuss the following optical properties of colloids :
- i) Visibility ii) Colour iii) Tyndall effects. (06 Marks)
- b. Write about the following electric properties of colloids :
- i) Electrophoresis (06 Marks)
 - ii) Electro osmosis. (04 Marks)
- c. Write a note on classification of emulsions with example.

OR

- 10 a. Explain the classification of colloids based on the nature of dispersed phase with example. (06 Marks)
- b. Explain about crystalloids and colloids. (04 Marks)
- c. Explain the following dynamic properties of colloids :
- i) Brownian movement
 - ii) Sedimentation
 - iii) Diffusion. (06 Marks)
